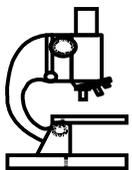


# Influence of Sulfate Source and Content on Hydration Kinetics and Compressive Strength of Portland Cement

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- Objectives of research
- Experimental results
  - heat of hydration
  - degree of hydration ( $C_3S$  and  $C_3A$ )
  - porosity, compressive strength
- Simulations
- Conclusions

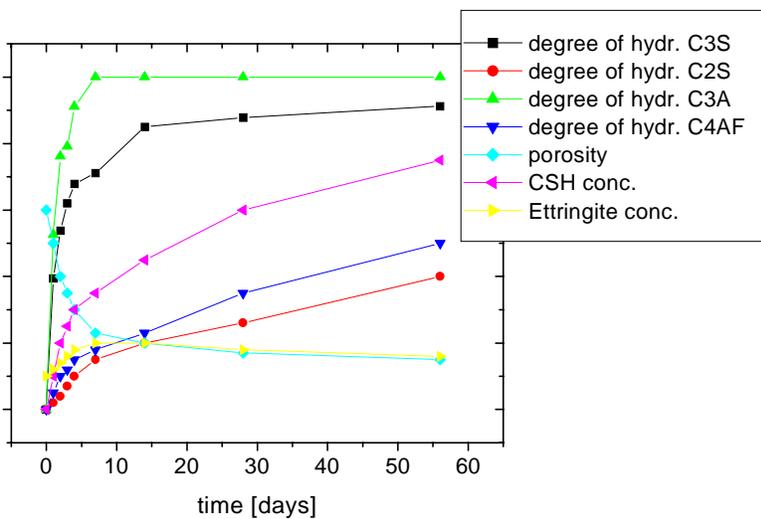


# Objectives of research: Predict compressive strength (of standard mortar; EN 196)

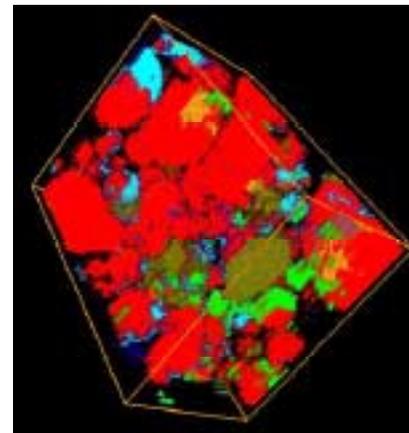
## 1<sup>st</sup> step: Simulate cement hydration

Characterize cement  
(bogue composition, gypsum, fly ash, silica fume,...)

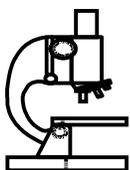
↓ NIST simulation software  
CEMHYD3D



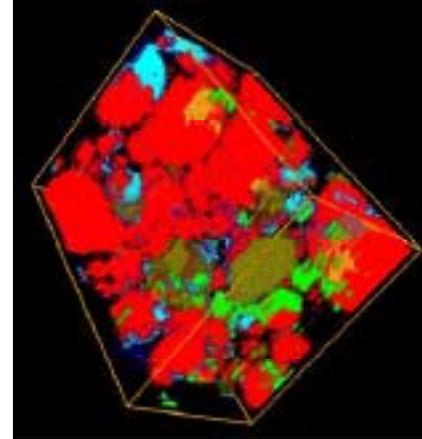
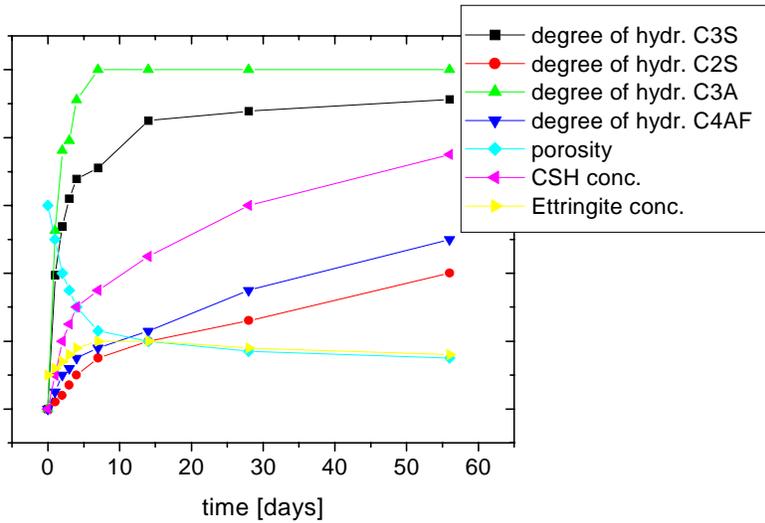
cement chemistry



3-D microstructure



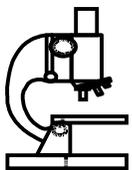
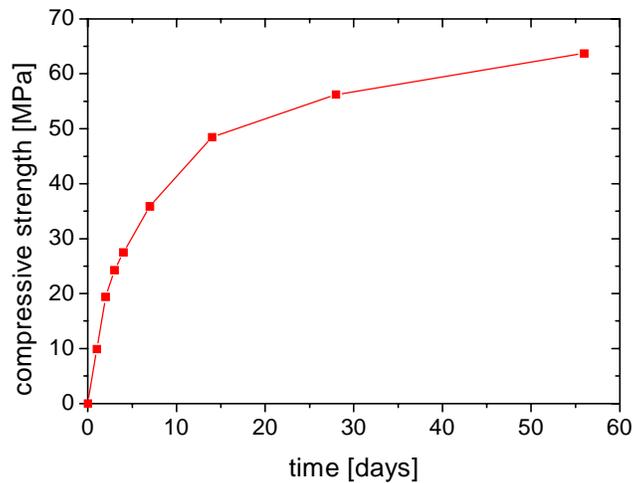
## 2<sup>nd</sup> step: calculate compressive strength



powers law:  $\sigma = \sigma_0 X^A$

$$X = \frac{0.68\alpha}{0.32\alpha + w/c}$$

calibration using N2



# Why does a cement producing company engage in modelling cement hydration?

## 1. Lower production costs

Quality control according to DIN 1164:

For every produced cement

- take two samples every week
  - test compressive strength after 2d, 7d, 28d
- ⇒ 312 tests/year for every cement

Reduce the number of tests

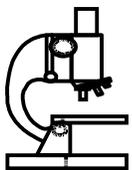
⇒ lower production costs

## 2. Speed up availability of information

Quality problems are evident after 28d

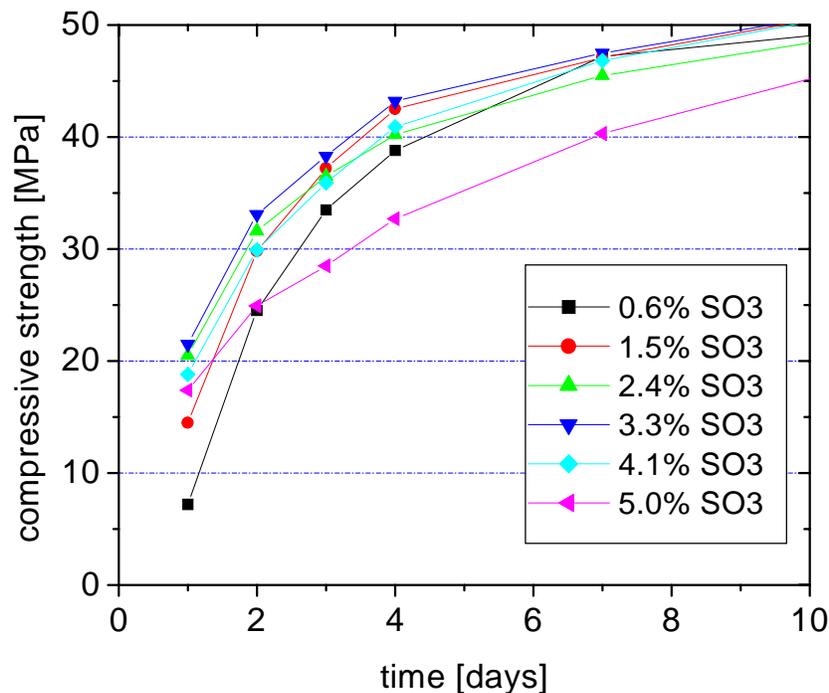
- what can be done if cement is already worked up?

In general simulated results are available after some h



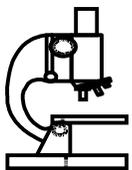
# Why have we investigated the influence of sulfate source on cement hydration?

**CaSO<sub>4</sub> highly influences development of compressive strength (+ cement hydration)**



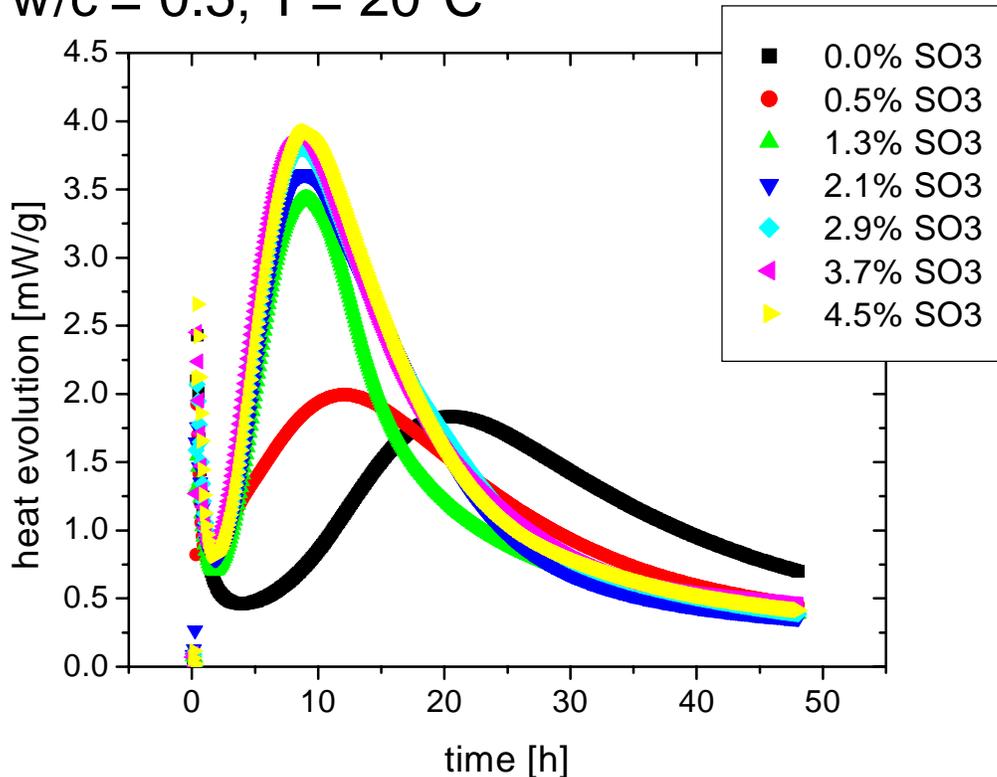
By our research we have tried

- to figure out the most important influences of CaSO<sub>4</sub> on the hydration kinetics of opc,
- to implement these results in CEMHYD3D and
- to develop a method to predict compr. strength for cements containing diff. amounts of CaSO<sub>4</sub>

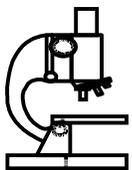


## Experimental results

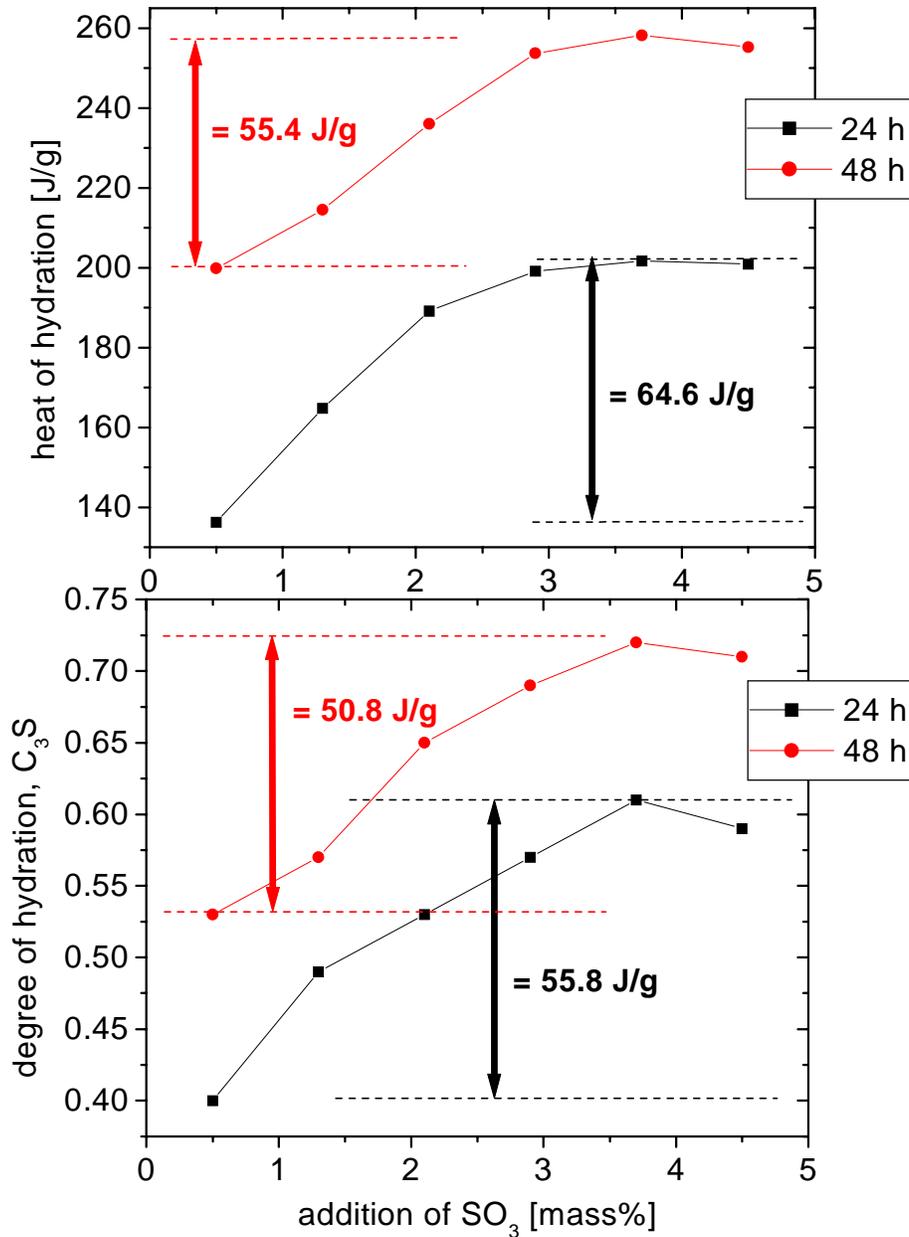
Ordinary portland clinker (5100 cm<sup>2</sup>/g)  
+ different amounts of anhydrite (10700 cm<sup>2</sup>/g)  
- w/c = 0.5; T = 20°C



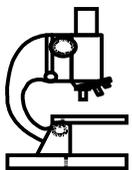
1. CaSO<sub>4</sub> accelerates cement hydration
2. A small amount of anhydrite (0.5% SO<sub>3</sub>) shortens the induction period, maximum value is equal
3. Increasing amounts of CaSO<sub>4</sub> further accelerate the rate of hydration



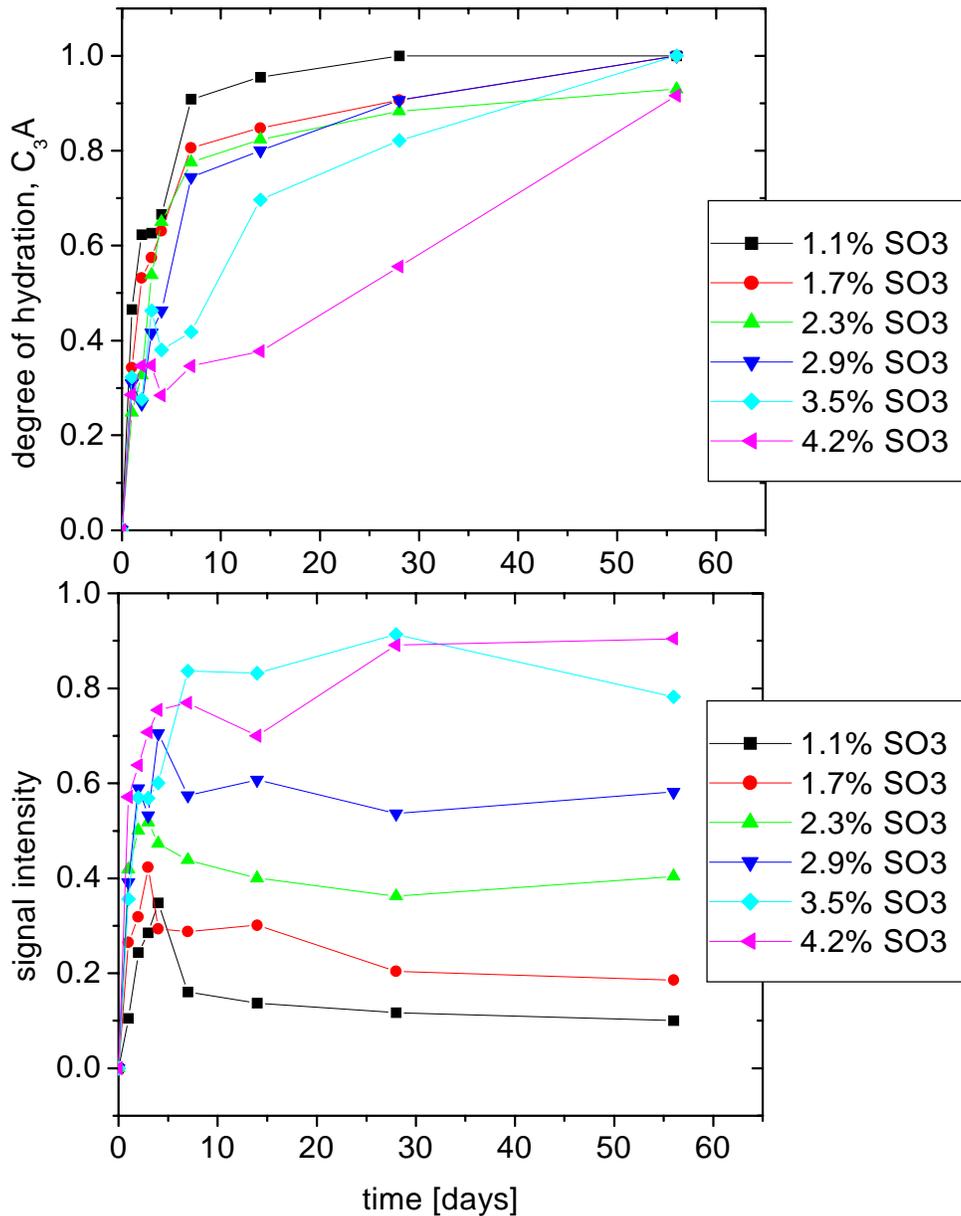
# Which reactions are responsible for the accelerated heat evolution?



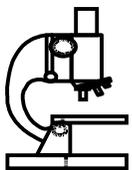
80% -90% of the acceleration can be attributed to the C<sub>3</sub>S reaction



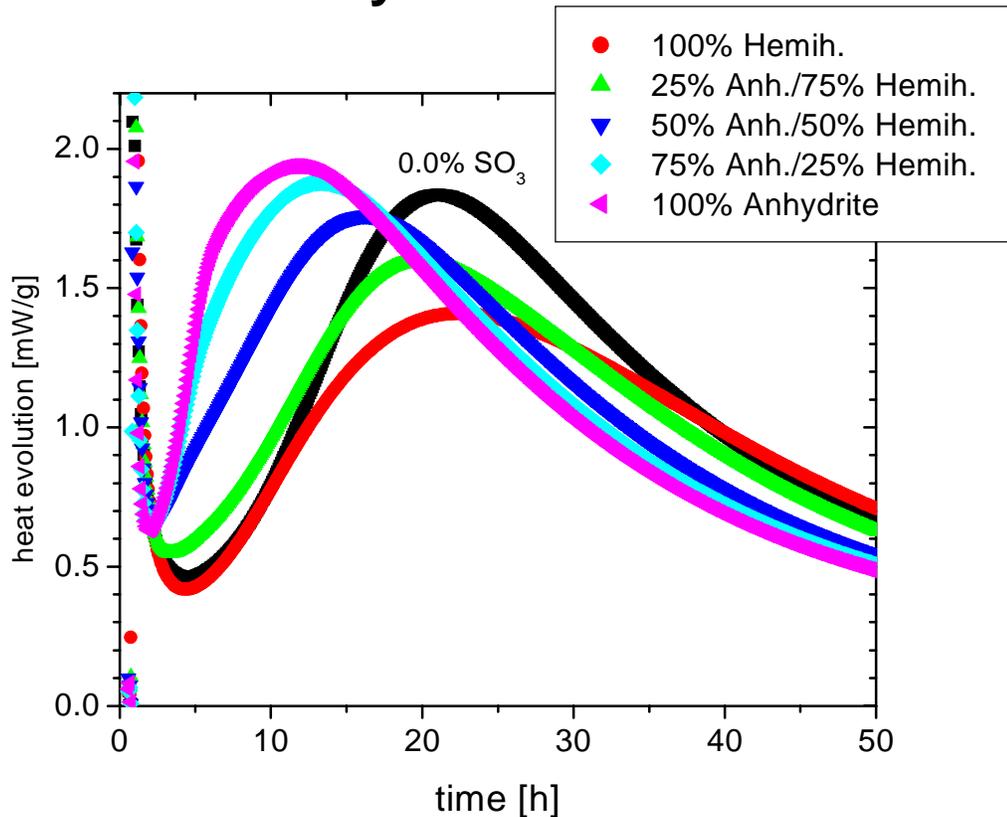
# Different amounts of $\text{CaSO}_4$ - $\text{C}_3\text{A}$ reactivity?



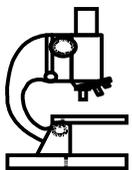
- High amounts of  $\text{CaSO}_4$  retard the hydration of  $\text{C}_3\text{A}$  for more than 14 days
- Despite retardation more ettringite is formed



# How do different modifications of $\text{CaSO}_4$ influence cement hydration?



- 1. Differences between hemihydrate and anhydrite**  $\Rightarrow$  different solubility
- 2.  $\text{CaSO}_4$  retards**  
higher solubility of hemihydrate  $\Rightarrow$  more ettringite retardation due to layer of „ettringite“
- 3.  $\text{CaSO}_4$  accelerates**  
Nonat:  $[\text{Ca}^{2+}]$  governs rate of reaction of  $\text{C}_3\text{S}$   
High  $[\text{Ca}^{2+}] \Rightarrow$  high rate of reaction  
Anhydrite reacts more slowly  $\Rightarrow$  longer available

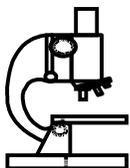
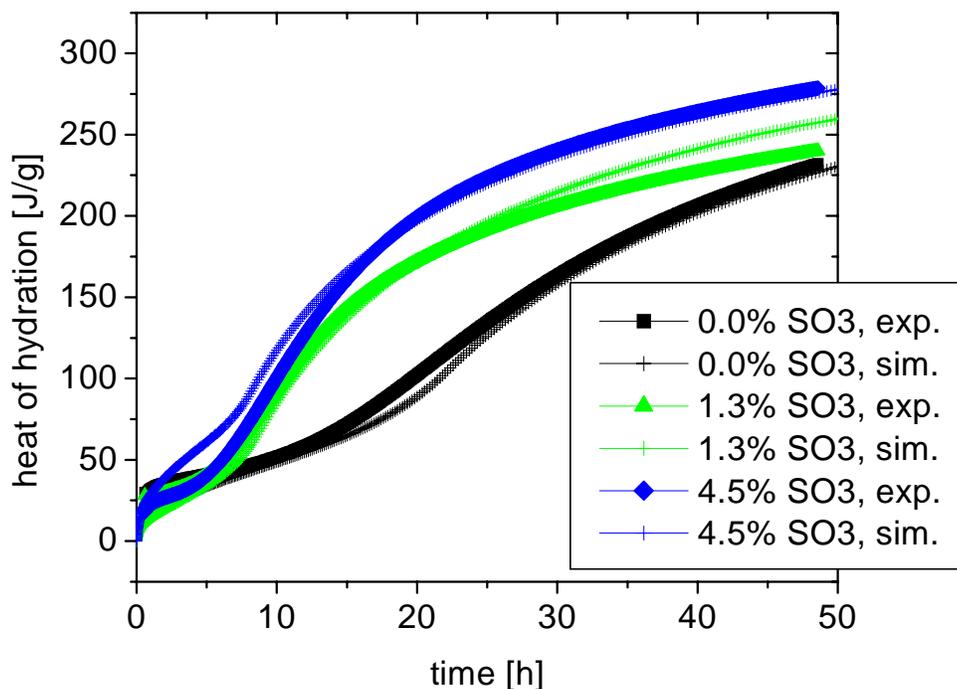


## First attempts to simulate the influence of $\text{CaSO}_4$ on cement hydration

measured isothermal heat evolution of  
- opc with different amounts of anhydrite  
-  $w/c = 0.5$

Simulations with NIST computer modelling software CEMHYD3D

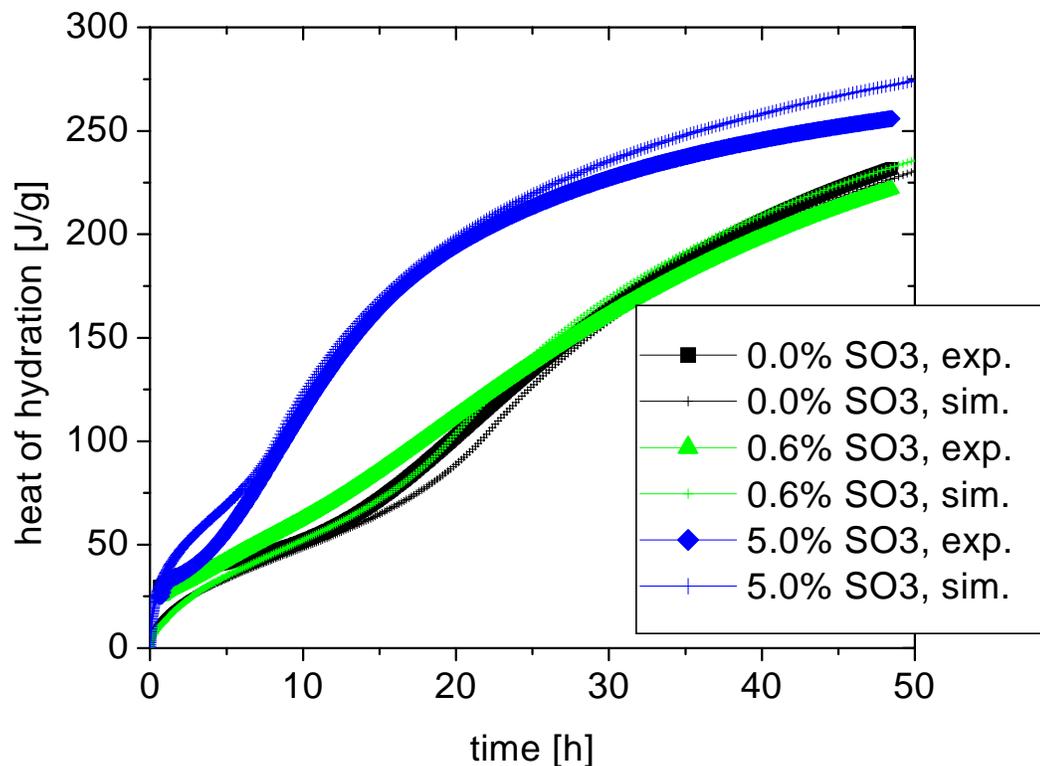
$$r_{\text{nuc}}(\text{CSH}) = f(c_{\text{CaSO}_4 \text{ in solution}})$$



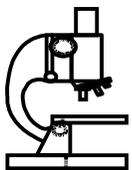
# Simulations on the influence of hemihydrate on cement hydration

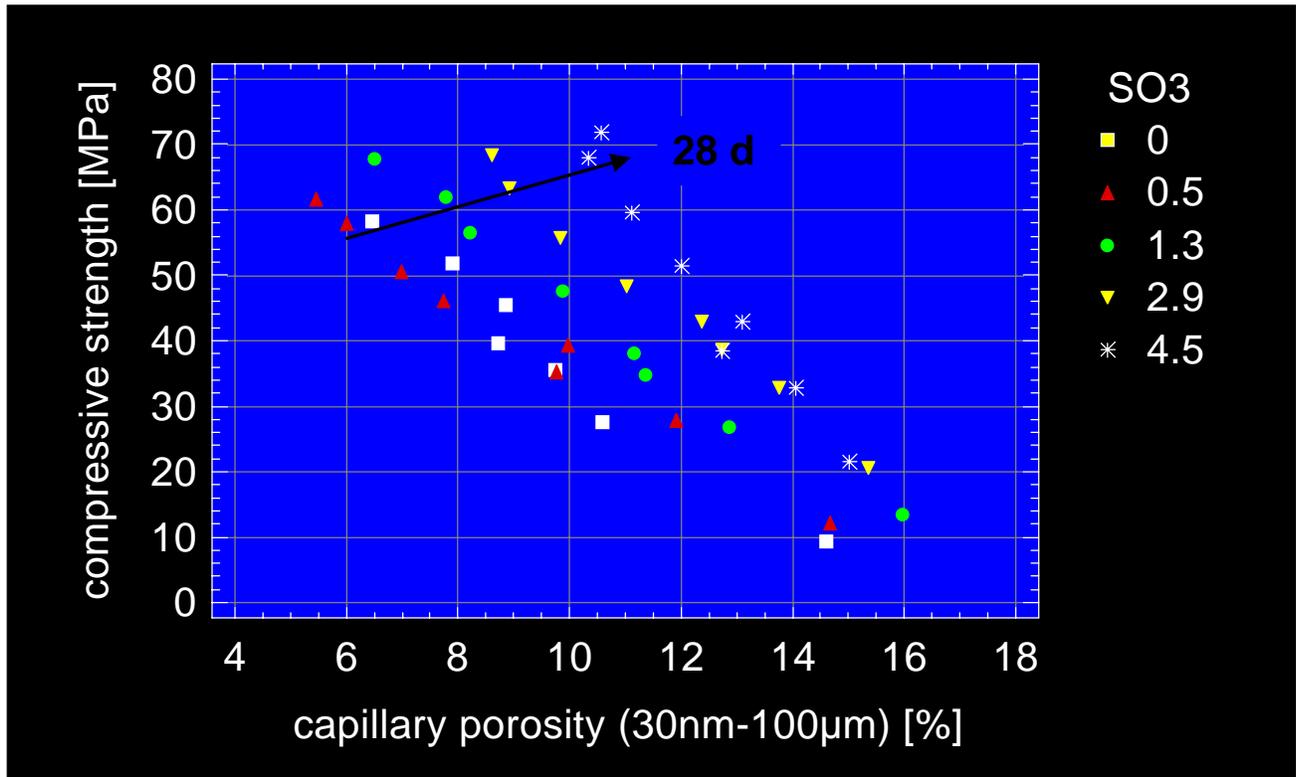
isothermal heat evolution of

- opc with different amounts of hemihydrate
- $w/c = 0.5$

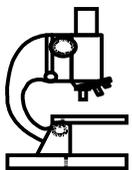


- Results look promising
- Model has still to be refined





- $\text{CaSO}_4$  modifies pore structure
- Poor overall correlation between strength and porosity
- Excellent correlation for each cement  $\Rightarrow$  calibrate each cement

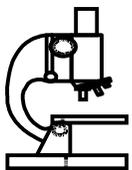
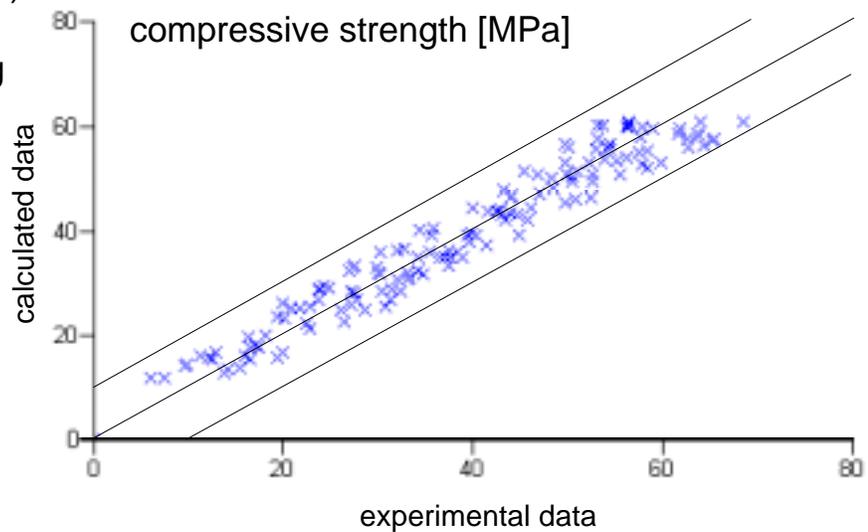


## Prediction of compressive strength

- Measured compressive strength according to EN 196
  - clinker composition (20 clinkers)
  - $\text{CaSO}_4$  content (0%-9%)
  - fineness (5 $\mu\text{m}$ -30 $\mu\text{m}$ )
- Simulation of cement hydration using CEMHYD3D
- Calibration of powers law using N2
- Predictions (1d - 56d)

maximum difference  
 $\approx 10 \text{ N/mm}^2$

average difference  
 $< 3.0 \text{ N/mm}^2$



# Conclusions

## Experimental results:

- $\text{CaSO}_4$  highly influences compressive strength
- $\text{CaSO}_4$  accelerates cement hydration
- The higher the amount of  $\text{CaSO}_4$ , the higher is the rate of hydration
- 80% - 90% of the acceleration can be attributed to the acceleration of the  $\text{C}_3\text{S}$  hydration
- $\text{CaSO}_4$  retards the hydration of  $\text{C}_3\text{A}$
- $\text{CaSO}_4$  increases capillary porosity + compressive strength

## Simulations:

- First attempts to simulate the influence of  $\text{CaSO}_4$  on cement hydration look promising
- Prediction of compressive strength is possible using CEMHYD3D and powers law (calibrated)  
Maximum deviation = 10 MPa  
Average deviation < 3.0 MPa

